

### 2008 Sunrise Free Radical School Presentation by: Balaraman Kalyanaraman, Ph.D.

### Evaluating Superoxide/Hydrogen Peroxide Detection Methods in Cells and Tissues – A Chemist's View



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> SFRBM 2008 Sunrise Free Radical School Saturday, November 22, 2008

### "Being able to do any chemistry was like being the one-eyed man in the kingdom of the blind."

- Roger Tsien

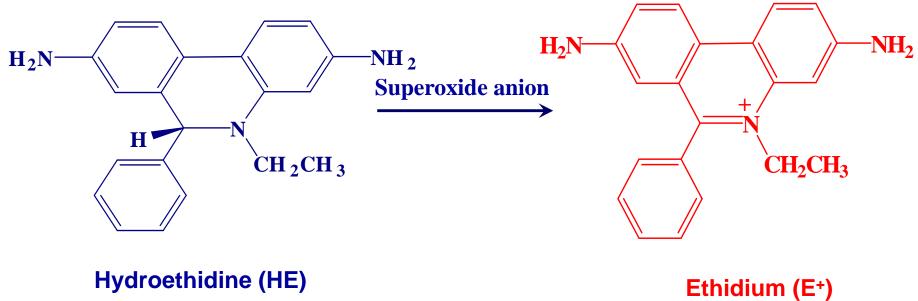
### Acknowledgements

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- Jeannette Vasquez Vivar (MCW)
- Hongtao Zhao (MCW, NIH)
- Jacek Zielonka (MCW)

# Intracellular detection of oxidants using fluorescent probes

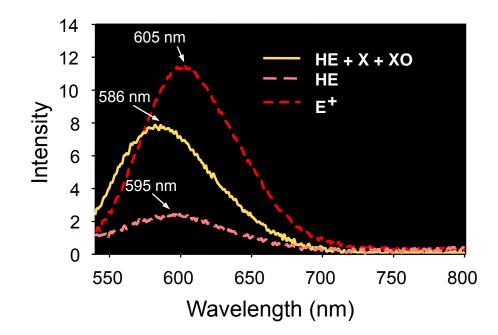
- The chemical reaction between the fluorescence probes and oxidants
- Redox reactions of fluorescent probe-derived radicals
- The cellular uptake and transport of probes
- Intracellular levels of reductants
- Translocation of fluorescent products in cells
- Understanding the chemistry and redox signaling mechanisms of inhibitors (apocynin, MnTBAP, Nacetylcysteine, etc.) that are frequently used to determine the nature of oxidants responsible for intracelluar fluorescence

### **Reaction between hydroethidine and superoxide**



(red fluorescence)

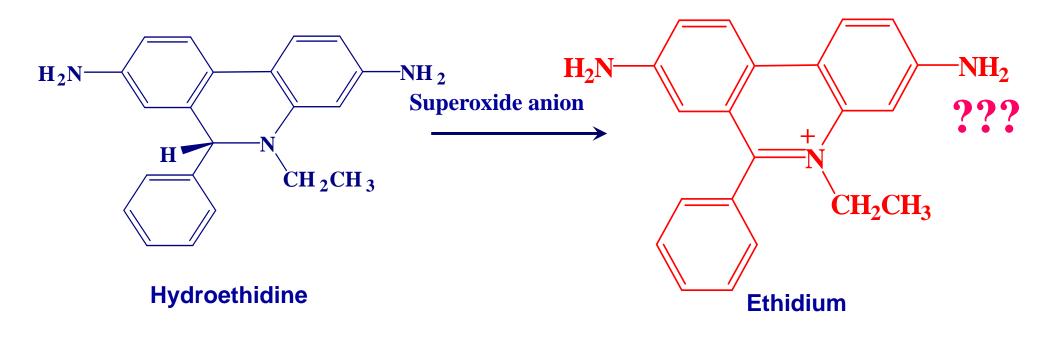
# Fluorescence spectra of the product formed during oxidation of HE by X/XO



Incubation mixtures consisted of HE (50  $\mu$ M), xanthine (1 mM), DTPA (100  $\mu$ M), and XO (0.05 U ml<sup>-1</sup>) in aerated phosphate buffer (100 mM, pH 7.4). Reaction was initiated by the addition of XO and the fluorescence spectrum recorded after 30 min. Shown for comparison is the spectrum obtained from E<sup>+</sup> (50  $\mu$ M) obtained under identical experimental conditions.

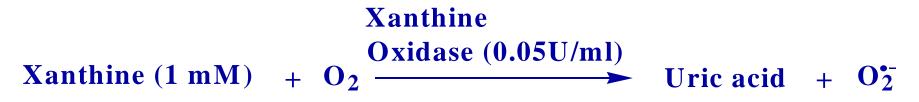
Zhao H, et al. FRBM 34:1359-1368, 2003

# Does superoxide react with hydroethidine to form ethidium?

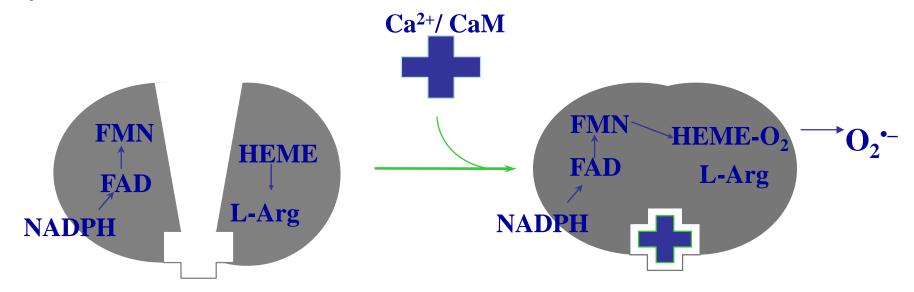


### Superoxide generating systems

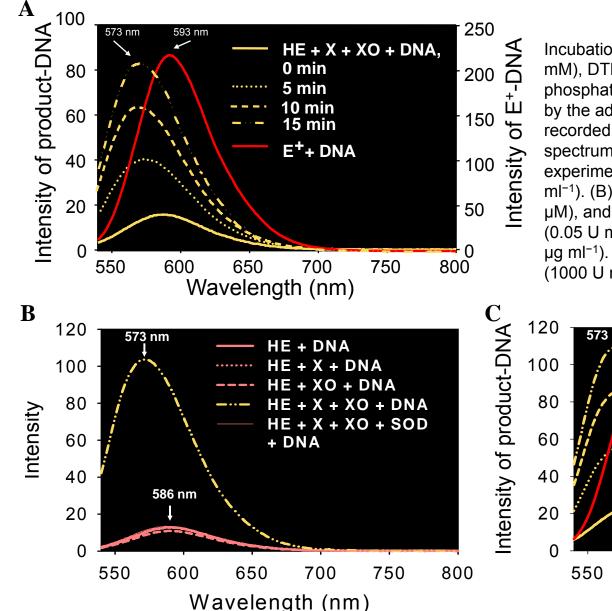
1. Xanthine/xanthine oxidase systems



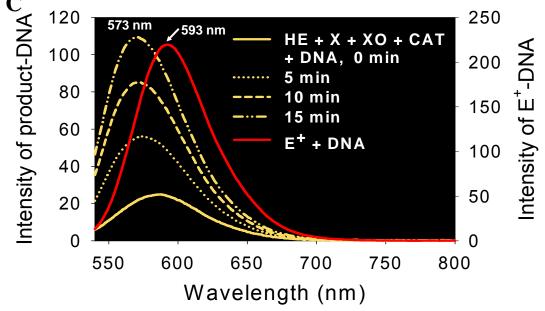
- 2. Potassium superoxide (KO<sub>2</sub>) in DMSO with crown ether
- 3. BH<sub>4</sub>-depleted eNOS systems



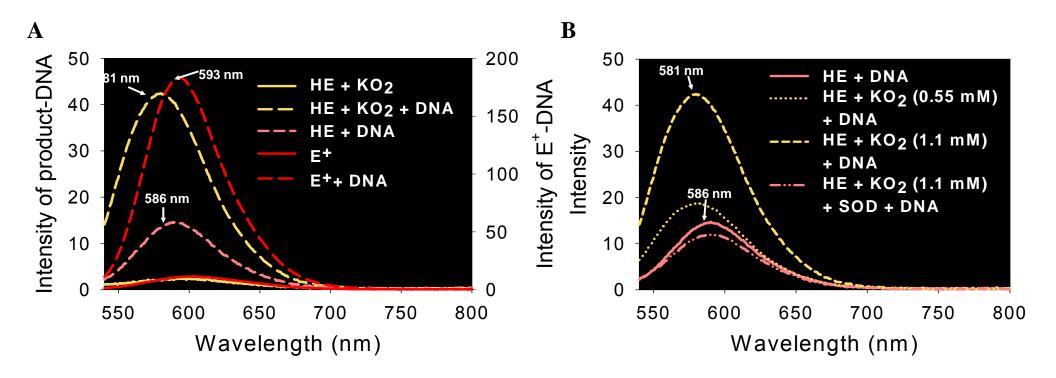
### Fluorescence spectra of HE/X/XO-DNA and EB-DNA



Incubation mixtures consisted of HE (50  $\mu$ M), xanthine (1 mM), DTPA (100  $\mu$ M), and XO (0.05 U ml<sup>-1</sup>) in aerated phosphate buffer (100 mM, pH 7.4). Reaction was initiated by the addition of XO and the fluorescence spectrum recorded after 30 min. Shown for comparison is the spectrum obtained from E<sup>+</sup> (50  $\mu$ M) obtained under identical experimental conditions. (A) in the presence of DNA (250  $\mu$ g ml<sup>-1</sup>). (B) Incubations contained HE (50  $\mu$ M), DTPA (100  $\mu$ M), and DNA (250  $\mu$ g ml<sup>-1</sup>) and xanthine (1 mM) or XO (0.05 U ml<sup>-1</sup>), both in the presence and absence of SOD (20  $\mu$ g ml<sup>-1</sup>). (C) Same as (B) but in the presence of catalase (1000 U ml<sup>-1</sup>).

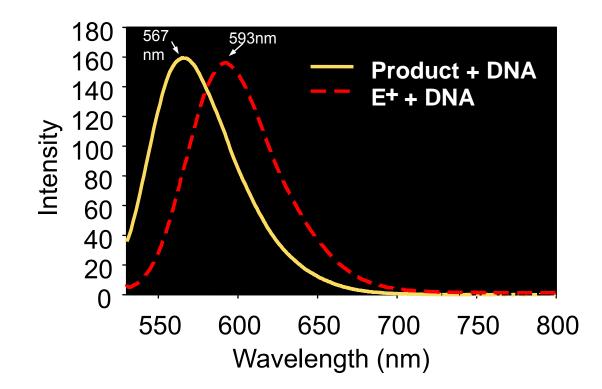


### Fluorescence spectra of HE/KO<sub>2</sub>-DNA and EB-DNA



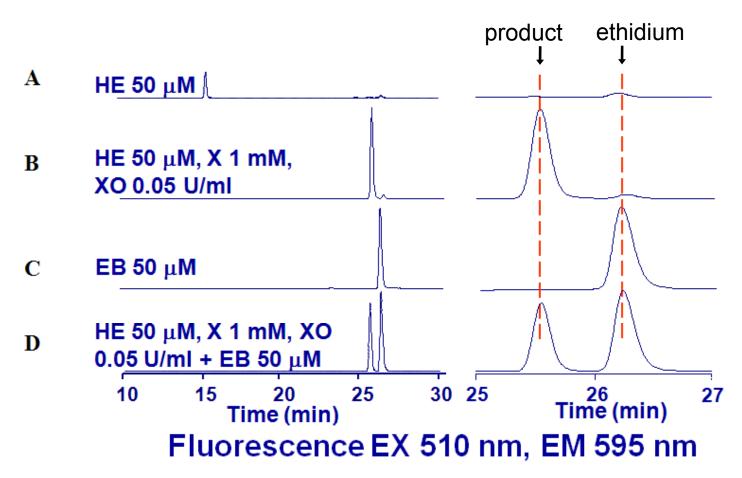
Fluorescence spectra of the product formed during oxidation of HE by potassium superoxide. (A) Incubations consisted of HE (50  $\mu$ M), DTPA (100  $\mu$ M), and KO<sub>2</sub> in the presence and absence of DNA (250  $\mu$ g ml<sup>-1</sup>) in phosphate buffer (100 mM, pH 7.4). Superoxide solution was prepared by adding 1 mg of KO<sub>2</sub> in 1 ml of dry DMSO and vortexed vigorously for 10 min. The reaction was initiated by adding 40  $\mu$ l of freshly prepared KO<sub>2</sub> in DMSO to 0.46 ml of the above buffer (final concentration of KO<sub>2</sub> was 1.1 mM). (B) Same as (A) but also containing SOD (20  $\mu$ g ml<sup>-1</sup>). The extent of oxidation of HE under these conditions is much less than that observed in X/XO.

# Fluorescence spectra of the product of HE reaction with $O_2^{-}$ and EB-DNA



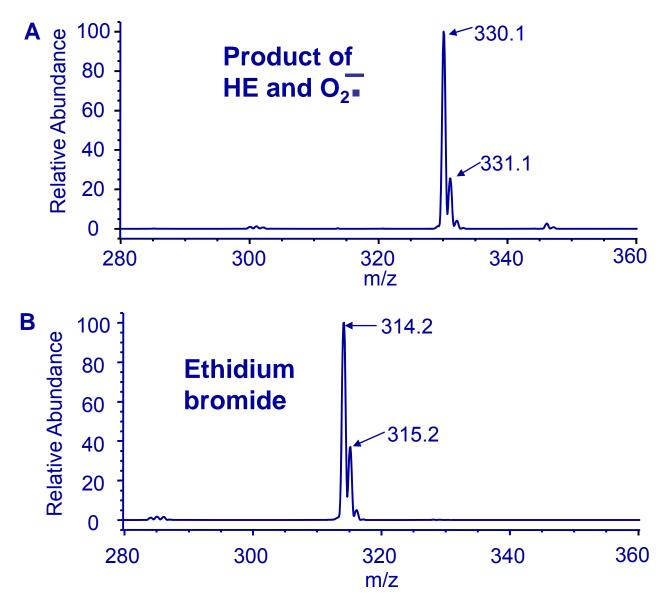
Fluorescence spectra of the purified product formed from oxidation of HE by X/XO and E<sup>+</sup> in the presence of DNA. Incubations contained 50  $\mu$ M of the pure product, 50  $\mu$ M E<sup>+</sup> and DNA (250  $\mu$ g ml<sup>-1</sup>) in phosphate buffer (100 mM, pH 7.4).

### **HPLC** analysis



HPLC chromatograms of HE, E<sup>+</sup>, and the product formed from oxidation of HE by the X/XO system. (A) Incubations contained xanthine (1 mM), HE (50  $\mu$ M), and DTPA (100  $\mu$ M). (B) Same as above but in the presence of XO (0.05 U ml<sup>-1</sup>). HPLC traces in (A) and (B) were obtained 30 min after starting the incubation. (C) HPLC trace of authentic E<sup>+</sup> (50  $\mu$ M) in phosphate buffer (100 mM, pH 7.4). (D) Same incubation conditions as in (B) but spiked with authentic E<sup>+</sup> (50  $\mu$ M). Right trace (A–D), HPLC chromatograms recorded on an expanded scale. Fluorescence detection at 510 nm (excitation) and 595 nm (emission) was used to monitor HE, E<sup>+</sup>, and the oxidation product of HE.

### Mass spectrometry



HPLC-mass spectrometry. (A) The mass spectrum of new product taken from the LC peak at retention time of 34.30 min; incubations contained xanthine (1 mM), HE (50 µM), and DTPA (100  $\mu$ M) and XO (0.05 U ml<sup>-1</sup>) in phosphate buffer (100 mM, pH 7.4). (B) The mass spectrum of E<sup>+</sup> taken from the LC peak at retention time of 33.97 min: incubations contained E<sup>+</sup>  $(50 \ \mu M)$  in phosphate buffer (100 mM, pH 7.4).

### Summary

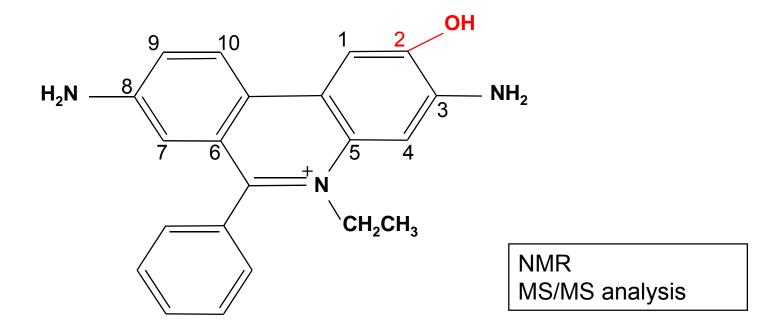
1. Hydroethidine reacts with superoxide to form a fluorescent compound, which is different from ethidium bromide. This compound is a specific product for hydroethidine reaction with superoxide. The fluorescence of the new compound is also be enhanced by DNA.

$$\mathsf{HE} + \mathsf{O}_2^- \longrightarrow \mathsf{E}^+$$

HE + O<sub>2</sub> - Characteristic fluorescent product that binds to DNA

- 2. Other oxidants (peroxynitrite, HOCI, H<sub>2</sub>O<sub>2</sub>, hydroxyl radical and peroxyl radicals) react with HE but do not form this particular product.
- 3. The new compound has a different retention time and molecular weight compared to ethidium cation.
- 4. The fluorescence properties of the new compound are different from that of ethidium bromide.

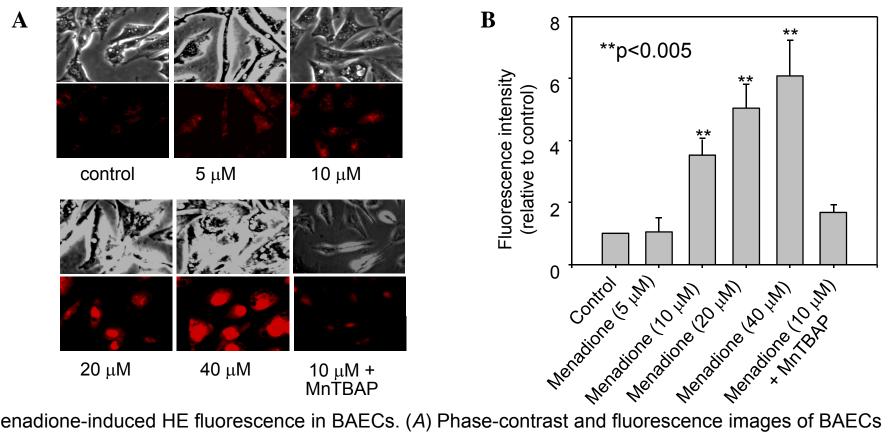
### Structure of the $HE/O_2^{-1}$ reaction product



### 2-Hydroxyethidium (2-OH-E+)

Zhao H, et al. PNAS 102:5727-32, 2005

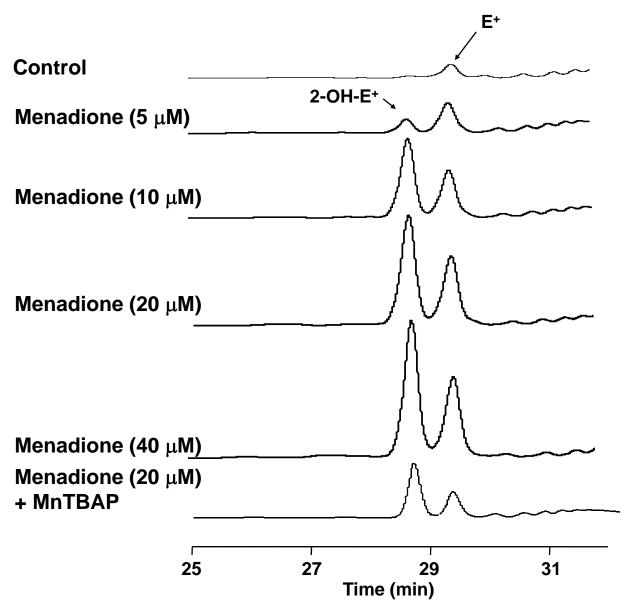
# Intracellular oxidation of HE: Menadione-induced HE fluorescence in BAEC



Menadione-induced HE fluorescence in BAECs. (*A*) Phase-contrast and fluorescence images of BAECs treated with different concentrations of menadione for 30 min. After menadione treatment, BAECs were washed with DPBS and incubated with HE (10  $\mu$ M) for 20 min. Cells were washed twice with DPBS and kept in the culture medium. The red fluorescence generated from HE was monitored by using fluorescence microscopy. BAECs were preincubated with MnTBAP (150  $\mu$ M) for 2 h. After washing the cells free of extracellular MnTBAP, they were treated with menadione (10  $\mu$ M) for 30 min, and HE-derived fluorescence was measured. (*B*) Densitometric analysis of data shown in *A*.

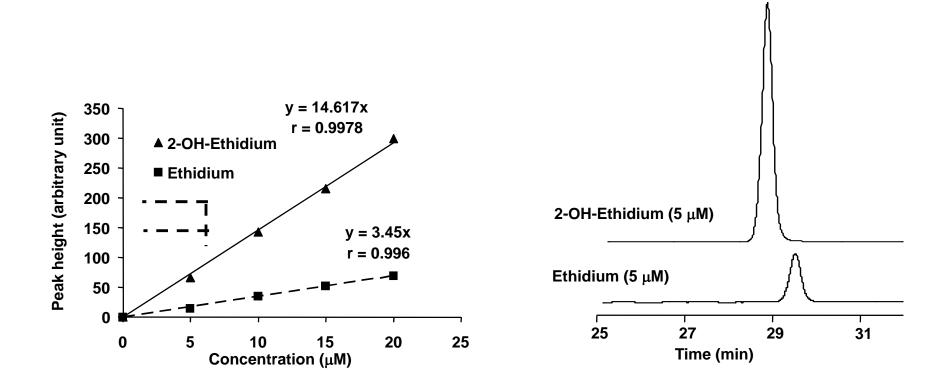
Zhao H, et al. PNAS 102:5727-32, 2005

# HPLC identification of 2-OH-E<sup>+</sup> from the O<sub>2</sub>-/HE reaction in BAECs

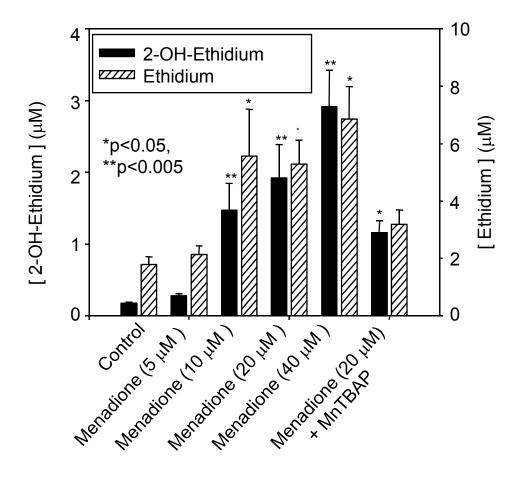


HPLC/fluorescence chromatograms of authentic 2-OH-E<sup>+</sup> (labeled 1) and E<sup>+</sup> (labeled 2). (*Inset*) HPLC peak intensity of 2-OH-E<sup>+</sup> and E<sup>+</sup> at different concentrations. BAECs were treated with various concentrations of menadione

### HPLC calibration using authentic compounds

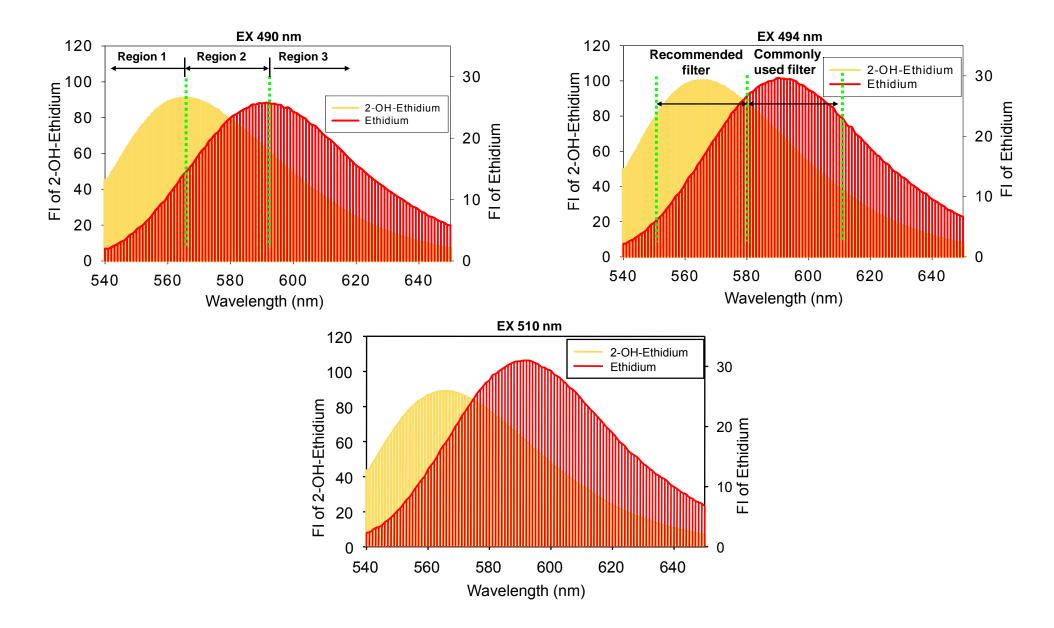


### Intracellular concentration of 2-OH-E<sup>+</sup> and E<sup>+</sup>



The actual concentrations of 2-OH-E<sup>+</sup> and E<sup>+</sup> in cell lysates generated under the conditions described above were calculated by using the calibration data shown

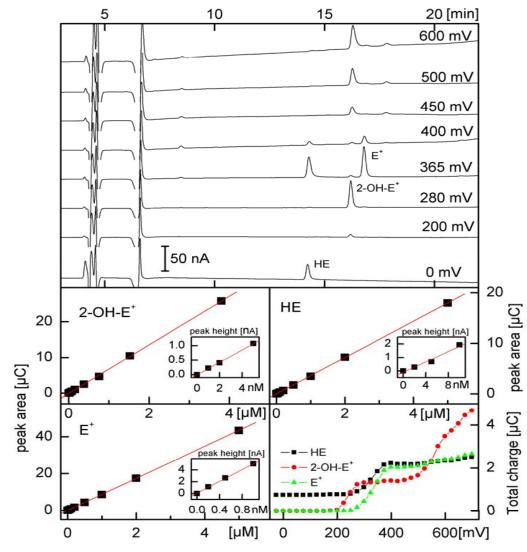
### **Optimal conditions for detection by fluorescence**



### Conclusions

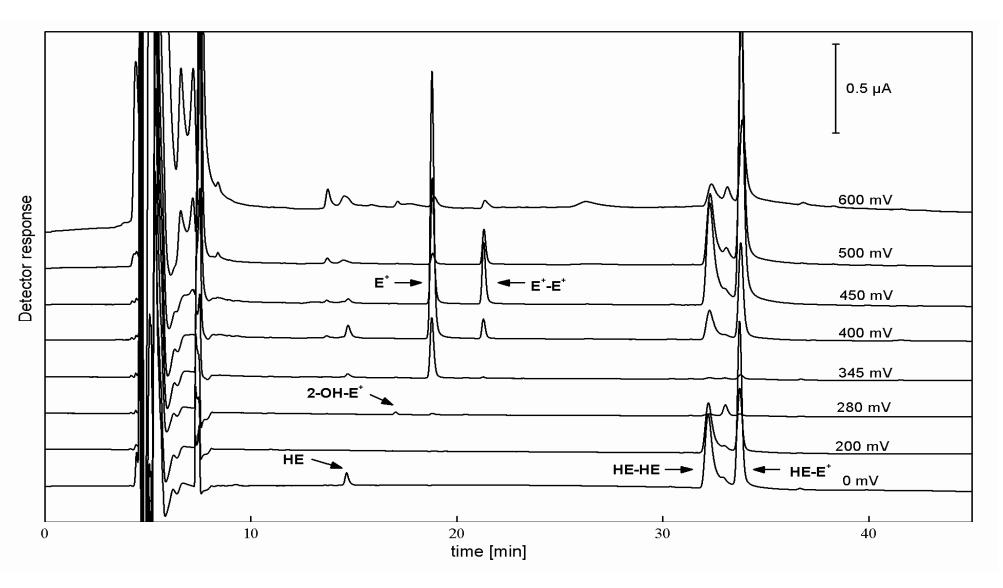
- Because of the overlapping fluorescence spectra from 2-OH-E<sup>+</sup> and E<sup>+</sup>, the "red fluorescence" formed from HE cannot be used to quantitate intracellular O<sub>2</sub>·formation.
- The new HPLC/fluorescence assay using HE as a probe is more suitable for quantifying intracellular O<sub>2</sub><sup>.-</sup>.
- Intracellular formation of 2-OH-E<sup>+</sup> is a diagnostic marker product of O<sub>2</sub>·<sup>-</sup>.

### HPLC/EC detection of HE, E<sup>+</sup>, and 2-OH-E<sup>+</sup>: A more sensitive methodology

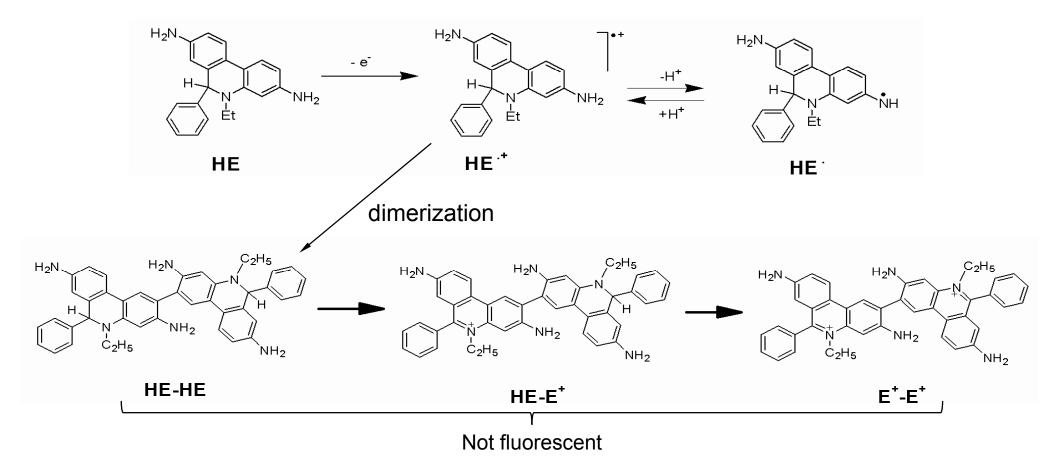


Zielonka J, et al. FRBM 41:1050-7, 2006

### **HPLC-EC** identification of dimers



### **Radical-mediated oxidation of HE**

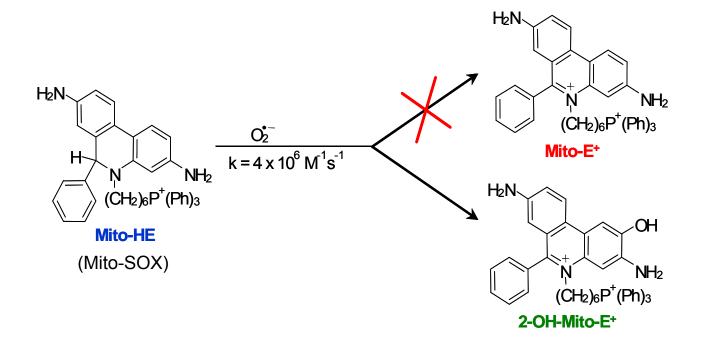


Zielonka J, et al. FRBM 44:835-46, 2008

# Geometry of the dimer molecule

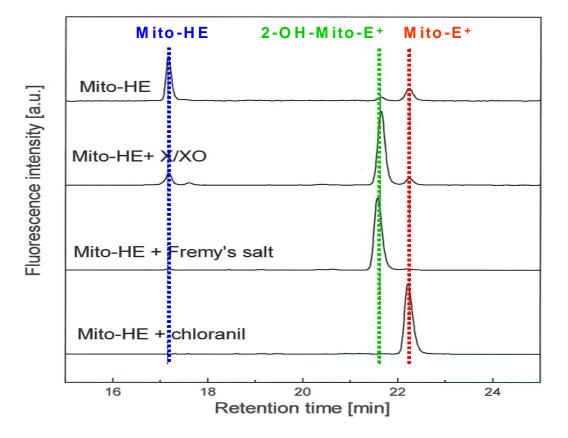
Zielonka J, et al. FRBM 44:835-46, 2008

# Superoxide reaction with Mito-HE leads to formation of a specific product, 2-OH-Mito-E+



Robinson KM, et al. *PNAS* 103:15038-43, 2006 Zielonka J, et al. *FRBM* 44:835-46, 2008

### HPLC-fluorescence detection of products formed from oxidation of Mito-HE

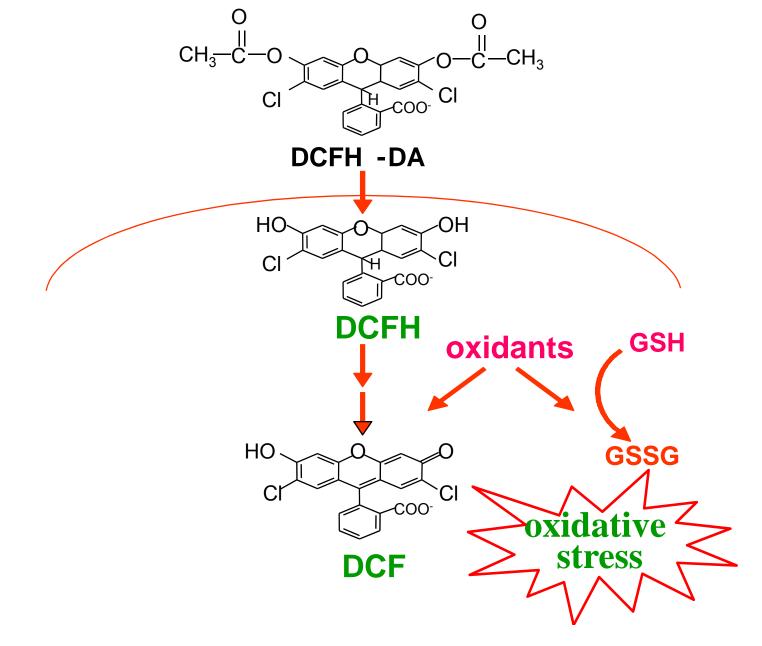


Zielonka J, et al. FRBM 44:835-46, 2008

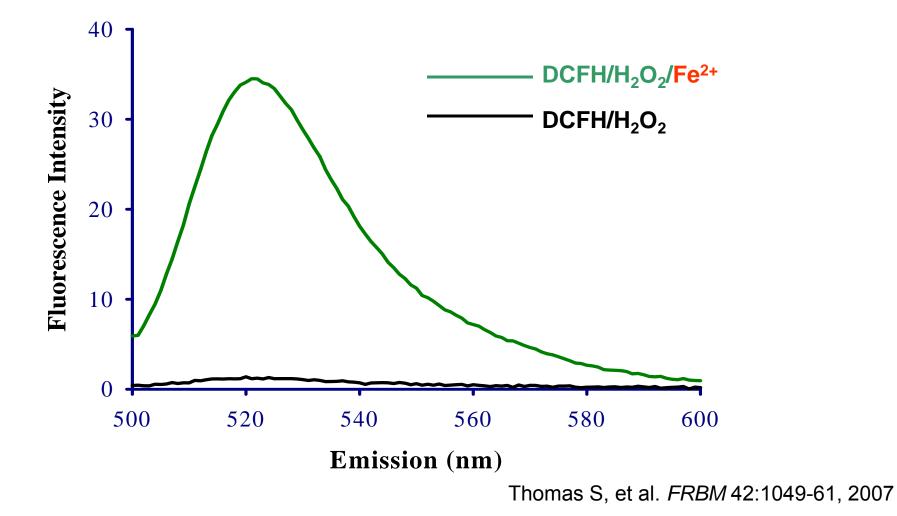
### **Overall conclusions**

- HE and its mitochondria-targeted analogs react with superoxide to form hydroxylated products that can be easily identified using HPLC techniques.
- However, prospects for exclusive visualization of these products in cells using fluorescence are not very good because of confounding oxidation chemistry of HE and Mito-HE.

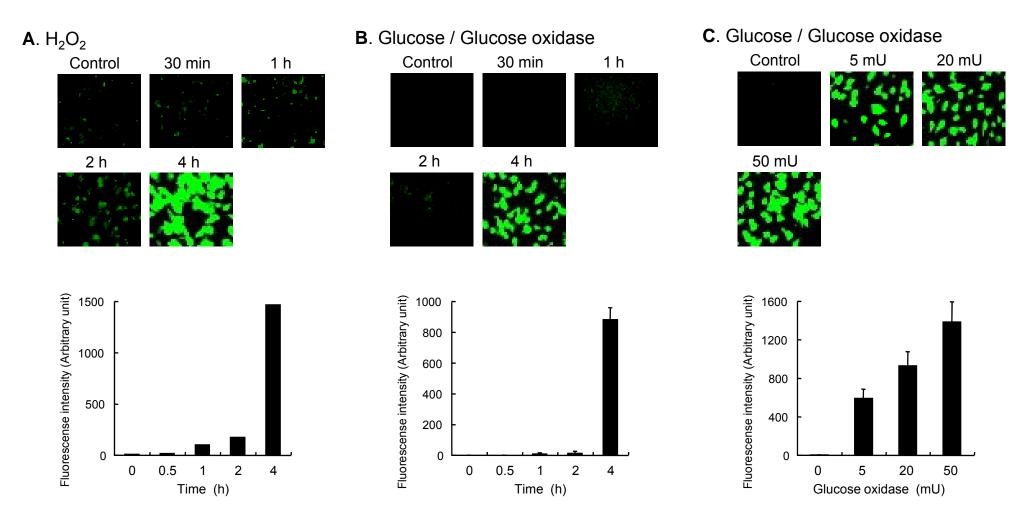
### Measurement of intracellular oxidative stress by DCF fluorescence



### Oxidation of DCFH by H<sub>2</sub>O<sub>2</sub> and iron

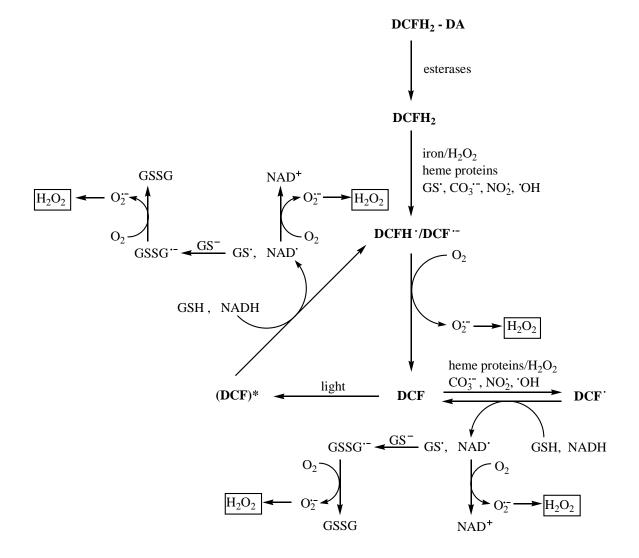


### Intracellular oxidation of DCFH to DCF



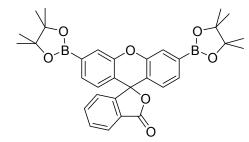
Tampo Y, et al. Circ Res 92:56-63, 2003

### **Redox reactions of DCF radical**

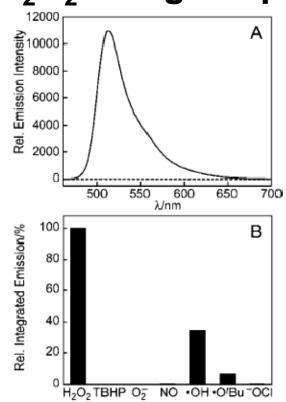


Zielonka J, et al. FRBM (in press, 2008) [doi:10.1016/j.freeradbiomed.2008.07.025]

### Intracellular detection of H<sub>2</sub>O<sub>2</sub> using the probe, PF1



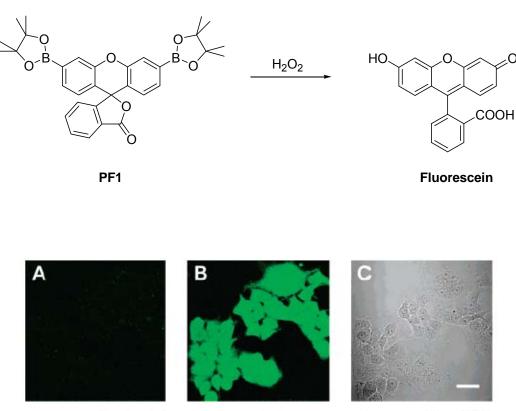




(A) Fluorescence response of 5  $\mu$ M PF1 to 100  $\mu$ M H<sub>2</sub>O<sub>2</sub>. The dotted and solid line spectra were recorded before and after H<sub>2</sub>O<sub>2</sub> addition, respectively. Spectra were acquired in 20 mM HEPES, pH 7 ( $\lambda_{exc} = 450$  nm). (B) Fluorescence responses of 5  $\mu$ M PF1 to various ROS (10 mM O<sub>2</sub><sup>-</sup>, 100  $\mu$ M for all other ROS). •OH and •O'Bu were generated by reaction of Fe<sup>2+</sup> with H<sub>2</sub>O<sub>2</sub> or *tert*-butyl hydroperoxide (TBHP), respectively. NO was delivered using *S*-nitrosocysteine (SNOC). Spectra were acquired in 20 mM HEPES, pH 7, and all data were obtained after incubation with the appropriate ROS at 25 °C for 1 h. Collected emission was integrated between 460 and 700 nm ( $\lambda_{exc} = 450$  nm).

Chang MCY, et al. J Am Chem Soc 126:15392-15393, 2004

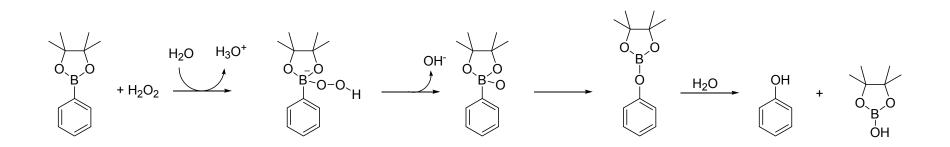
### Intracellular detection of H<sub>2</sub>O<sub>2</sub>



Confocal fluorescence and phase contrast images of live HEK cells. (A) Fluorescence image of HEK cells incubated with 5  $\mu$ M PF1 for 5 min at 25 °C. (B) Fluorescence image of PF1-stained HEK cells treated with 100  $\mu$ M H<sub>2</sub>O<sub>2</sub> for 5 min at 25 °C. (C) Brightfield image of live HEK cells after H<sub>2</sub>O<sub>2</sub> addition to confirm viability. Scale bar = 30  $\mu$ m.

Chang MCY, et al. J Am Chem Soc 126:15392-15393, 2004

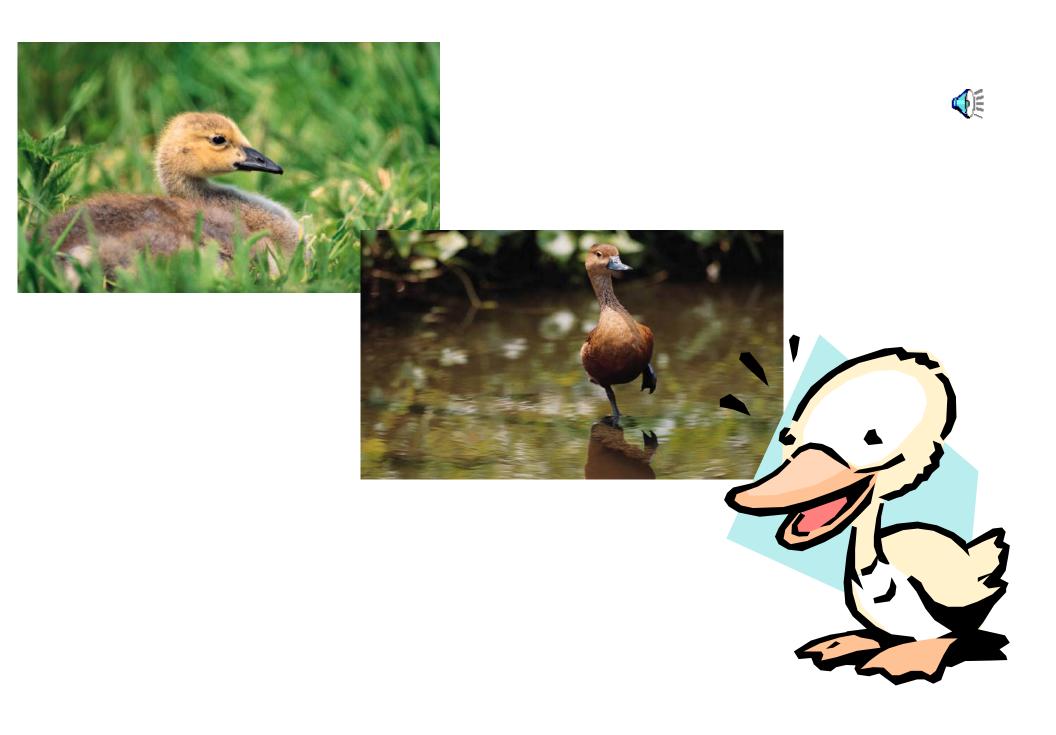
### **Back to the drawing board!**



Based on: Kuivila HG. J. Am. Chem. Soc. 76:870-874, 1954

### **Relevant publications**

- Zhao H, Kalivendi S, Zhang H, Joseph J, Nithipatikom K, Vasquez-Vivar J, Kalyanaraman B. Superoxide reacts with hydroethidine but forms a fluorescent product that is distinctly different from ethidium: potential implications in intracellular fluorescence detection of superoxide. *Free Radic Biol Med* 34:1359-1368, 2003.
- Zhao H, Joseph J, Fales HM, Sokoloski EA, Levine RL, Vasquez-Vivar J, Kalyanaraman B. Detection and characterization of the product of hydroethidine and intracellular superoxide by HPLC and limitations of fluorescence. *Proc Natl Acad Sci USA* 102:5727-5732, 2005. [Erratum: *Proc Natl Acad Sci USA* 102:9086, 2005.]
- Zielonka J, Zhao H, Xu Y, Kalyanaraman B. Mechanistic similarities between oxidation at hydroethidine by Fremy's salt and superoxide: stopped-flow optical and EPR studies. *Free Radic Biol Med* 39:853-863, 2005.
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- Dhanasekaran A, Kotamraju S, Karunakaran C, Kalivendi SV, Thomas S, Joseph J, Kalyanaraman B. Mitochondria superoxide dismutase mimetic inhibits peroxide-induced oxidative damage and apoptosis: role of mitochondrial superoxide. *Free Radic Biol Med* 39:567-83, 2005.
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- Zielonka J, Sarna T, Roberts JE, Wishart JF, Kalyanaraman B. Pulse radiolysis and steady-state analyses of the reaction between hydroethidine and superoxide and other oxidants. *Arch Biochem Biophys* 456:39-47, 2006.
- Thomas S, Kotamraju S, Zielonka J, Harder DR, Kalyanaraman B Hydrogen peroxide induces nitric oxide and proteasome activity in endothelial cells: a bell-shaped signaling response. *Free Radic Biol Med* 42:1049-61, 2007.
- Zielonka J, Vasquez-Vivar J, Kalyanaraman B. Detection of 2-hydroxyethidium in cellular systems: a unique marker product of superoxide and hydroethidine. *Nature Protocols* 3:8-21, 2008.
- Zielonka J, Srinivasan S, Hardy M, Ouari O, Lopez M, Vasquez-Vivar J, Avadhani NG, Kalyanaraman B. Cytochrome c-mediated oxidation of hydroethidine and mito-hydroethidine in mitochondria: identification of homo- and heterodimers. *Free Radic Biol Med* 44:835-46, 2008.
- Zielonka J, Kalyanaraman B. ROS-generating mitochondrial DNA mutations can regulate tumor cell metastasis A critical commentary. *Free Radic Biol Med* (in press, 2008). [doi:10.1016/j.freeradbiomed.2008.07.025]





## Happy Thanksgiving!



